

# Enantioenriched Bifunctional Crotylsilanes for the Asymmetric Synthesis of Orthogonally Protected 2-Methyl-1,3-diols

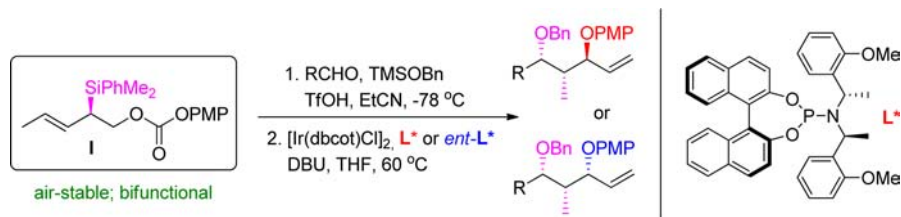
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## ABSTRACT



Enantiomerically pure  $\alpha$ -substituted crotylsilane reagents **I** and *ent*-**I** undergo asymmetric aldehyde crotylation followed by Ir(I)-catalyzed diastereoselective allylic etherification to give a variety of orthogonally protected 2-methyl-1,3-diols at the synthetically useful level of yields and stereoselectivity. The reagents are air-stable and bifunctional so that they can be used in these reactions sequentially without recourse to functional group adjustments.

Enantioenriched  $\alpha$ -substituted allyl/crotylsilanes have played an important role in organic synthesis.<sup>1</sup> Many

synthetically useful structures can be accessed through the use of these reagents.<sup>2</sup> Furthermore, numerous total syntheses of biologically and therapeutically important compounds have been accomplished by employing these reagents in the key step.<sup>3</sup> As such, developing new such reagents and/or novel synthetic strategies based on these reagents has been a focus of recent studies.<sup>4</sup>

Numerous studies, mainly from Panek's group, have established that acid-mediated asymmetric crotylation of aldehydes and acetals by chiral  $\alpha$ -substituted crotylsilanes

(1) (a) Masse, C. E.; Panek, J. S. *Chem. Rev.* **1995**, *95*, 1293–1316. (b) Fleming, I.; Barbero, A.; Walter, D. *Chem. Rev.* **1997**, *97*, 2063–2192. (c) Chabaud, L.; James, P.; Landais, Y. *Eur. J. Org. Chem.* **2004**, 3173–3199. (d) Bates, R. H.; Chen, M.; Roush, W. R. *Curr. Opin. Drug Discovery Dev.* **2008**, *11*, 778–792. (e) Shimizu, M.; Hiyama, T. *Handbook of Functionalized Organometallics* **2005**, *1*, 173–202.

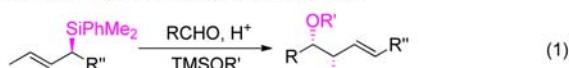
(2) (a) Panek, J. S.; Yang, M. *J. Am. Chem. Soc.* **1991**, *113*, 9868–9870. (b) Micalizio, G. C.; Roush, W. R. *Org. Lett.* **2000**, *2*, 461–464. (c) Peng, Z.-H.; Woerpel, K. A. *Org. Lett.* **2001**, *3*, 675–678. (d) Huang, H.; Panek, J. S. *J. Am. Chem. Soc.* **2000**, *122*, 9836–9837. (e) Su, Q.; Panek, J. S. *J. Am. Chem. Soc.* **2004**, *126*, 2425–2430. (f) Su, Q.; Panek, J. S. *Angew. Chem., Int. Ed.* **2005**, *44*, 1223–1225. (g) Binanzer, M.; Fang, G. Y.; Aggarwal, V. K. *Angew. Chem., Int. Ed.* **2010**, *49*, 4264–4268. (h) Wu, J.; Zhu, K.-C.; Yuan, P.-W.; Panek, J. S. *Org. Lett.* **2012**, *14*, 3624–3627. (i) Wu, J.; Becerril, J.; Lian, Y.; Davies, H. M. L.; Porco, J. A., Jr.; Panek, J. S. *Angew. Chem., Int. Ed.* **2011**, *50*, 5938–5942. (j) Franz, A. K.; Dreyfuss, P. D.; Schreiber, S. L. *J. Am. Chem. Soc.* **2007**, *129*, 1020–1021. (k) Lipomi, D. J.; Panek, J. S. *Org. Lett.* **2005**, *7*, 4701–4704.

(3) (a) Wu, J.; Panek, J. S. *J. Org. Chem.* **2011**, *76*, 9900–9918. (b) Wu, J.; Panek, J. S. *Angew. Chem., Int. Ed.* **2010**, *49*, 6165–6168. (c) Robinson, A.; Aggarwal, V. K. *Angew. Chem., Int. Ed.* **2010**, *49*, 6673–6675. (d) Youngsaye, W.; Lowe, J. T.; Pohlki, F.; Ralifo, P.; Panek, J. S. *Angew. Chem., Int. Ed.* **2007**, *46*, 9211–9214. (e) Va, P.; Roush, W. R. *J. Am. Chem. Soc.* **2006**, *128*, 15960–15961. (f) Wrona, I. E.; Gabarda, A. E.; Evano, G.; Panek, J. S. *J. Am. Chem. Soc.* **2005**, *127*, 15026–15027. (g) Tinsley, J. M.; Roush, W. R. *J. Am. Chem. Soc.* **2005**, *127*, 10818–10819. (h) Su, Q.; Panek, J. S. *J. Am. Chem. Soc.* **2004**, *126*, 2425–2430.

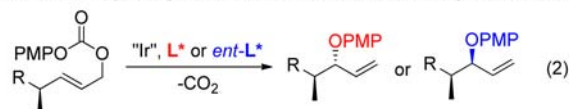
(4) (a) Nagao, K.; Yokobori, U.; Makida, Y.; Ohmiya, H.; Sawamura, M. *J. Am. Chem. Soc.* **2012**, *134*, 8982–8987. (b) Hodgson, D. M.; Salik, S. *Org. Lett.* **2012**, *14*, 4402–4405. (c) Chen, M.; Roush, W. R. *Org. Lett.* **2011**, *13*, 1992–1995. (d) Aggarwal, V. K.; Binanzer, M.; de Ceglie, M. C.; Gallanti, M.; Glasspoole, B. W.; Kendrick, S. J. F.; Sonawane, R. P.; Vázquez-Romero, A.; Webster, M. P. *Org. Lett.* **2011**, *13*, 1490–1493. (e) Selander, N.; Paasch, J. R.; Szabó, K. J. *J. Am. Chem. Soc.* **2011**, *133*, 409–411. (f) Han, S. B.; Gao, X.; Krische, M. J. *J. Am. Chem. Soc.* **2010**, *132*, 9153–9156. (g) Fukuda, K.; Miyashita, M.; Tanino, K. *Tetrahedron Lett.* **2010**, *51*, 4523–4525. (h) Li, D.; Tanaka, T.; Ohmiya, H.; Sawamura, M. *Org. Lett.* **2010**, *12*, 3h44–3347. (i) Hayashi, S.; Hirano, K.; Yorimitsu, H.; Oshima, K. *J. Am. Chem. Soc.* **2007**, *129*, 12650–12651. (j) Lira, R.; Roush, W. R. *Org. Lett.* **2007**, *9*, 4315–4318. (k) Kacprzynski, M. A.; May, T. L.; Kazane, S. A.; Hoveyda, A. H. *Angew. Chem., Int. Ed.* **2007**, *46*, 4554–4558. (l) Vyas, D. J.; Oestreich, M. *Angew. Chem., Int. Ed.* **2010**, *49*, 8513–8515 (racemic version).

in the presence of TMSOR' can give rise to *E*-homoallylic ethers with high enantio- and diastereoselectivities (eq 1).<sup>1a,5</sup> However, the resulting homoallylic ethers have been used primarily for simple functional group interconversions and have rarely been employed in catalytic asymmetric reactions where a new bond is stereoselectively introduced by using a pair of enantiomeric catalysts/ligands. This may be in large part due to the fact that almost all prior chiral  $\alpha$ -substituted crotylsilanes, to our knowledge, produce homoallylic ethers, which are not appropriately appended with functional groups for subsequent catalytic asymmetric reactions.<sup>6</sup> In this regard, new chiral  $\alpha$ -substituted crotylsilanes that can undergo an asymmetric crotylation immediately followed by a catalytic asymmetric reaction without recourse to functional group adjustments would be highly desirable. We considered this need in relation with our recent reports that described Ir(I)-catalyzed

Prior work: asymmetric aldehyde crotylation



Prior work: Ir(I)-catalyzed double diastereoselective allylic etherification

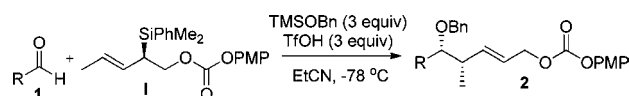









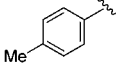

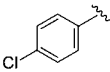
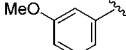
The current work: merging the above two reactions into 1



diastereoselective allylic etherification of *p*-methoxyphenyl (PMP) allyl carbonates (eq 2)<sup>7,8</sup> and wondered whether it would be possible to merge Panek's asymmetric aldehyde crotylation with our Ir(I)-catalyzed diastereoselective allylic etherification to devise a new chiral bifunctional crotylsilane **I** (eq 3). Herein we report that the crotylsilane **I** and its enantiomer *ent*-**I** indeed undergo the above two reactions sequentially to deliver orthogonally protected 2-methyl-1,3-diols in a highly stereoselective and efficient fashion, which are common structural motifs in many natural products.<sup>9</sup> We believe that the

**Table 1.** Asymmetric Crotylation of Aldehydes by the Crotylsilanes **I** and *ent*-**I**



entry	R-		yield [%] <sup>a</sup>	syn:anti <sup>b</sup>	ee [%] <sup>c</sup>
1		(a)	92	4:1	97
2		(b)	93	7:1	96
3		(c)	88	>25:1	97
4		(d)	82	>25:1	98
5		(e)	91	>25:1	94
6		(f)	93	>25:1	97
7		(g)	92	>25:1	96
8		(h)	94	>25:1	96
9		(i)	94	>25:1	97
10		(j)	96	>25:1	95
11		(k)	88	>25:1	95

<sup>a</sup> Isolated yields. <sup>b</sup> Determined by the <sup>1</sup>H NMR spectrum of reaction mixtures. <sup>c</sup> Determined by chiral HPLC.

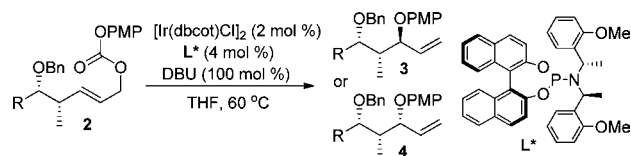
developed synthetic strategy utilizing **I** and *ent*-**I** are compared very favorably with other state-of-art methods for the asymmetric synthesis of 2-methyl-1,3-diol derivatives.<sup>9</sup>

The enantioenriched crotylsilanes **I** and *ent*-**I** were prepared by acylating the corresponding known alcohols<sup>5</sup> with ClCO<sub>2</sub>PMP in the presence of pyridine in methylene

- (5) Wu, J.; Chen, Y.; Panek, J. S. *Org. Lett.* **2010**, *12*, 2112–2115.  
 (6) (a) Wu, J.; Zhu, K.-C.; Yuan, P.-W.; Panek, J. S. *Org. Lett.* **2012**, *14*, 3624–3627. (b) Panek, J. S.; Yang, M.; Solomon, J. S. *J. Org. Chem.* **1993**, *58*, 1003–1010. In these papers, crotylation products were used in Pd(II)-catalyzed intramolecular allylic acetate transposition and Pd(0)-catalyzed intramolecular allylic substitution reactions. Since these reactions are intramolecular reactions and their stereochemistry is controlled by substrate stereochemistry or solvents used, they are not true catalytic asymmetric reactions, where a new bond is stereoselectively introduced by a pair of enantiomeric catalysts/ligands.  
 (7) (a) Kim, D.; Reddy, S.; Singh, O. V.; Lee, J. S.; Kong, S. B.; Han, H. *Org. Lett.* **2013**, *15*, 512–515. (b) Kim, D.; Lee, J. S.; Kong, S. B.; Han, H. *Angew. Chem., Int. Ed.* **2013**, *52*, 4203–4206.  
 (8) For reviews on Ir(I)-catalyzed allylation: (a) Hartwig, J. F.; Pouy, M. J. *Top. Organomet. Chem.* **2011**, *34*, 169–208. (b) Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. *Chem. Commun.* **2007**, 675–691.

- (9) For recent approaches, see: (a) Gao, X.; Han, H.; Krische, M. J. *J. Am. Chem. Soc.* **2011**, *133*, 12795–12800. (b) Ichibakase, T.; Nakajima, M. *Org. Lett.* **2011**, *13*, 1579–1581. (c) Rohr, K.; Herre, R.; Mahrwald, R. *J. Org. Chem.* **2009**, *74*, 3744–3749. (d) Mlynarski, J.; Rakiel, B.; Stodulski, M.; Suszczyńska, A.; Frelek, J. *Chem.—Eur. J.* **2006**, *12*, 8158–8167. (e) Gnanadesikan, V.; Horiuchi, Y.; Oshima, T.; Shibasaki, M. *J. Am. Chem. Soc.* **2004**, *126*, 7782–7783. (f) Mascarenhas, C. M.; Miller, S. P.; White, P. S.; Morken, J. P. *Angew. Chem., Int. Ed.* **2001**, *40*, 601–603. (g) Evans, D. A.; Hoveyda, A. H. *J. Am. Chem. Soc.* **1990**, *112*, 6447–6449. (h) Dorgan, P. D.; Durrani, J.; Cases-Thomas, M. J.; Hulme, A. N. *J. Org. Chem.* **2010**, *75*, 7475–7478. (i) Esteve, J.; Matas, S.; Pellicena, M.; Velasco, J.; Romea, P.; Urpi, F.; Font-Bardia, M. *Eur. J. Org. Chem.* **2010**, 3146–3151. (j) Zbieg, J. R.; Yamaguchi, E.; McInturff, E. L.; Krische, M. J. *Science* **2012**, *336*, 324–327. (k) McInturff, E. L.; Yamaguchi, E.; Krische, M. J. *J. Am. Chem. Soc.* **2012**, *134*, 20628–20631. (l) Kim, I. S.; Han, S. B.; Krische, M. J. *J. Am. Chem. Soc.* **2009**, *131*, 2514–2520. (m) Gao, X.; Townsend, I. A.; Krische, M. J. *J. Org. Chem.* **2011**, *76*, 2350–2354.

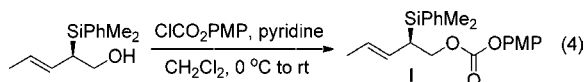
**Table 2.** Ir(I)-Catalyzed Decarboxylative Double Diastereoselective Allylic Etherification of PMP Allyl Carbonates **2** in the Presence of Phosphoramidite Ligand **L\***<sup>a</sup>



entry	R-	product	yield [%] <sup>b</sup>	3:4 dr <sup>c</sup>
1		(1 <i>S</i> , 2 <i>R</i> , 3 <i>S</i> )- <b>3b</b>	81	10:1
2		(1 <i>S</i> , 2 <i>R</i> , 3 <i>S</i> )- <b>3c</b>	75	14:1
3		(1 <i>S</i> , 2 <i>R</i> , 3 <i>S</i> )- <b>3d</b>	82	>25:1
4		(1 <i>S</i> , 2 <i>R</i> , 3 <i>S</i> )- <b>3e</b>	81	>25:1
5		(1 <i>R</i> , 2 <i>S</i> , 3 <i>S</i> )- <b>3f</b>	77	20:1
6		(1 <i>R</i> , 2 <i>S</i> , 3 <i>S</i> )- <b>3g</b>	74	>25:1
7		(1 <i>R</i> , 2 <i>S</i> , 3 <i>S</i> )- <b>3h</b>	83	21:1
8		(1 <i>R</i> , 2 <i>S</i> , 3 <i>S</i> )- <b>3i</b>	36 <sup>d</sup>	>25:1
9		(1 <i>R</i> , 2 <i>S</i> , 3 <i>S</i> )- <b>3j</b>	74	>25:1
10		(1 <i>R</i> , 2 <i>S</i> , 3 <i>S</i> )- <b>3k</b>	75	11:1

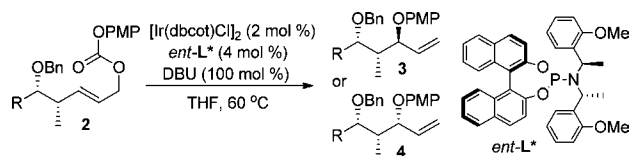
<sup>a</sup> PMP allyl carbonate (0.2 mmol), [Ir(dbcot)Cl]<sub>2</sub> (0.004 mmol), **L\*** (0.008 mmol), DBU (0.2 mmol) in THF (3 mL) at 60 °C. <sup>b</sup> Isolated yields. <sup>c</sup> Determined by the <sup>1</sup>H NMR spectrum of reaction mixtures. <sup>d</sup> Linear product/branched product = 1:1.

chloride (eq 4). The crotylsilanes **I** and *ent*-**I** are air-stable and can be stored without any precautions.



Acid-mediated asymmetric crotylation of benzaldehyde by **I** in the presence of TMSOBn was used to determine the optimal conditions. After various acids (TfOH, TMSOTf, TMSI, TiCl<sub>4</sub>, and BF<sub>3</sub>·OEt<sub>2</sub>), solvents (EtCN, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, and toluene), and temperatures (−78, −50, −20, and 0 °C) were screened, as well as the amounts of TMSOBn, the reaction conditions involving TfOH (3 equiv) and TMSOBn (3 equiv) in EtCN at −78 °C were determined to be optimal to give the corresponding homoallylic ether in 95% yield in 5 h. The conditions using 1.5 equiv of TfOH could be used, but the reaction became very sluggish, taking >24 h to complete. When less than 3 equiv of TMSOBn were used, the corresponding tetrahydrofuran

**Table 3.** Ir(I)-Catalyzed Decarboxylative Double Diastereoselective Allylic Etherification of PMP Allyl Carbonates **2** in the Presence of Phosphoramidite Ligand *ent*-**L\***<sup>a</sup>



entry	R-	product	yield [%] <sup>b</sup>	3:4 dr <sup>c</sup>
1		(1 <i>S</i> , 2 <i>R</i> , 3 <i>R</i> )- <b>4b</b>	84	1:>25
2		(1 <i>S</i> , 2 <i>R</i> , 3 <i>R</i> )- <b>4c</b>	79	1:>25
3		(1 <i>S</i> , 2 <i>R</i> , 3 <i>R</i> )- <b>4d</b>	86	1:>25
4		(1 <i>S</i> , 2 <i>R</i> , 3 <i>R</i> )- <b>4e</b>	84	1:>25
5		(1 <i>R</i> , 2 <i>S</i> , 3 <i>R</i> )- <b>4f</b>	82	1:16
6		(1 <i>R</i> , 2 <i>S</i> , 3 <i>R</i> )- <b>4g</b>	83	1:>25
7		(1 <i>R</i> , 2 <i>S</i> , 3 <i>R</i> )- <b>4h</b>	86	1:13
8		(1 <i>R</i> , 2 <i>S</i> , 3 <i>R</i> )- <b>4i</b>	41 <sup>d</sup>	1:10
9		(1 <i>R</i> , 2 <i>S</i> , 3 <i>R</i> )- <b>4j</b>	83	1:13
10		(1 <i>R</i> , 2 <i>S</i> , 3 <i>R</i> )- <b>4k</b>	80	1:13

<sup>a</sup> PMP allyl carbonate (0.2 mmol), [Ir(dbcot)Cl]<sub>2</sub> (0.004 mmol), *ent*-**L\*** (0.008 mmol), DBU (0.2 mmol) in THF (3 mL) at 60 °C. <sup>b</sup> Isolated yields. <sup>c</sup> Determined by the <sup>1</sup>H NMR spectrum of reaction mixtures. <sup>d</sup> Linear product/branched product = 1:1.

byproduct started to form.<sup>5,10</sup> The crotylsilane **I** tended to decompose to give PMPOH at higher temperatures than −20 °C. All attempts to generate the corresponding homoallylic alcohol were unsuccessful.

The optimal conditions were applied to a group of aldehydes, and the results are presented in Table 1. Unbranched linear aliphatic aldehydes (entries 1 and 2) resulted in high yields, but exhibited modest *syn/anti* selectivities (4–7:1). On the other hand, branched (entries 3 and 4) and cyclic (entry 5) aliphatic aldehydes worked well to give the corresponding homoallylic ethers in good yields and with excellent *syn/anti* selectivities (>25:1). Uniformly high yields (≥88%) and excellent *syn/anti* selectivities (>25:1) were obtained with all aromatic aldehydes tested (entries 6–11). The electron-donating and -withdrawing ability of a substituent on the benzene ring of aromatic

(10) Heitzman, C. L.; Lambert, W. T.; Mertz, E.; Shotwell, J. B.; Tinsley, J. M.; Va, P.; Roush, W. R. *Org. Lett.* **2005**, *7*, 2405–2408.

aldehydes, as well as its position, had very little influence on the diastereoselectivity and reaction yield. In all cases studied, excellent enantioselectivities ( $\geq 94\%$ ) were obtained, indicating that the reactions proceeded with near-perfect chirality transfer. The relative and absolute stereochemistry of **2** were assigned by hydrolyzing **2f** to the known allylic alcohol (see the Supporting Information) and analogy to the literature.<sup>5</sup>

Table 2 depicts Ir(I)-catalyzed decarboxylative double diastereoselective allylic etherification of **2** in the presence of phosphoramidite ligand **L\***, which gives rise to orthogonally protected 2-methyl-1,3-diols.<sup>11</sup> The catalytic conditions involving [Ir(abcot)Cl]<sub>2</sub> (2 mol %), **L\*** (4 mol %), and DBU (100 mol %) in THF that we recently reported for Ir(I)-catalyzed enantioselective decarboxylative allylic etherification were used at 60 °C.<sup>7</sup> Gratifyingly, high yields and diastereoselectivities were obtained with almost all aliphatic and aromatic substrates studied. Only the *o*-chlorophenyl substrate (entry 8) exhibited a modest yield (36%) due to the formation of the corresponding linear allylation product, but still showed excellent > 25:1 diastereoselectivity.

Ir(I)-catalyzed decarboxylative double diastereoselective allylic etherification of **2** was also conducted with phosphoramidite ligand *ent*-**L\*** under otherwise identical conditions, and the results are shown in Table 3. Again, high yields and diastereoselectivities were observed in all cases except for the *o*-chlorophenyl substrate (entry 8), which suffered from modest yield. Compared with *o*-tolyl and *p*-chlorophenyl substrates that gave a good branched to linear ratio and reaction yield, these results are surprising and the reason is not clear. The use of *ent*-**L\*** reversed

the diastereofacial preference to favor the other diastereomers **4**. The stereochemistry of the allylic etherification was assigned by converting **3c** and **4c** to the respective known compound (see the Supporting Information) and by analogy to the literature.<sup>8,12</sup> Taken together with those in Table 2, these stereochemical outcomes clearly indicate that double diastereoselectivity in the Ir(I)-catalyzed decarboxylative allylic etherification of **2** is primarily governed by ligand/catalyst stereochemistry (reagent-controlled stereochemistry).<sup>11</sup>

In summary, we report new bifunctional chiral crotylsilanes **I** and *ent*-**L\***, which are air-stable and can undergo highly enantio-/diastereoselective aldehyde crotylation followed by Ir(I)-catalyzed double diastereoselective decarboxylative allylic etherification. To the best of our knowledge, these reagents represent the first example of chiral  $\alpha$ -substituted crotylsilanes that can undergo asymmetric aldehyde crotylation followed by a catalytic asymmetric reaction without recourse to functional group adjustments. Considering the prevalence of 2-methyl-1,3-diol motifs in natural products and biologically interesting compounds, the method should find broad applications in organic synthesis.

**Acknowledgment.** Research reported in this publication was supported by the National Science Foundation (CHE 0911134).

**Supporting Information Available.** Experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(12) Welter, C.; Koch, O.; Lipowsky, G.; Helmchen, G. *Chem. Commun.* **2004**, 896–897.

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(11) Masamune, S.; Choy, W.; Petersen, J. S.; Sita, L. R. *Angew. Chem., Int. Ed.* **1985**, *24*, 1–30.